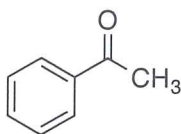
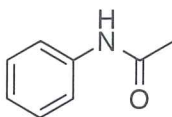


## 1. Nomenclature

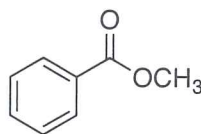
(a) (10 points) Each compound below contains an arene. List the other functional group in each compound.



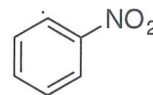
ketone



amide



ester

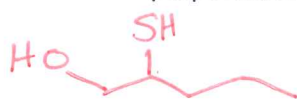


nitro

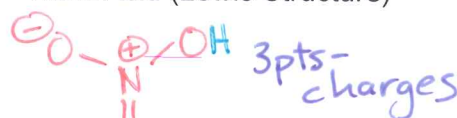
3pts each

(b) (20 points) Draw structures corresponding to the following names.

2-Mercaptopentanol

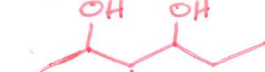


Nitric Acid (Lewis Structure)

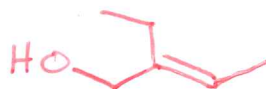


5pts each

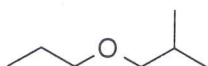
2,4-Hexanediol



(E)-2-Ethyl-2-buten-1-ol



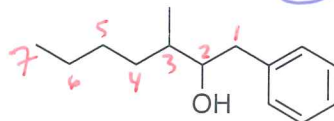
(c) (20 points) Provide IUPAC names for any three of the following compounds. "X" out the ones to skip.



Propyl (2-methylpropyl) ether

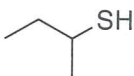
Propyl isobutyl ether

(2) (2) (2)



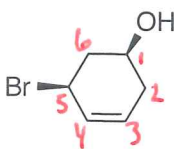
1-Phenyl-3-methyl-2-heptanol

(2) (2) (2)



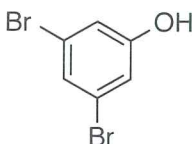
2-butanethiol or 2-mercaptobutane

(2) (2) (2)



(1S,5S)-5-Bromo-3-cyclohexenol

(1) (1) (1) (1) (1) (1)

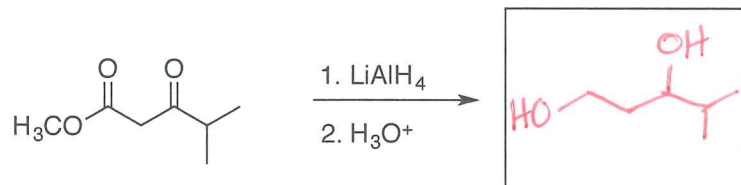
3,5-Dibromophenol  
meta-Dibromophenol

(2) (2) (2)

6pts each

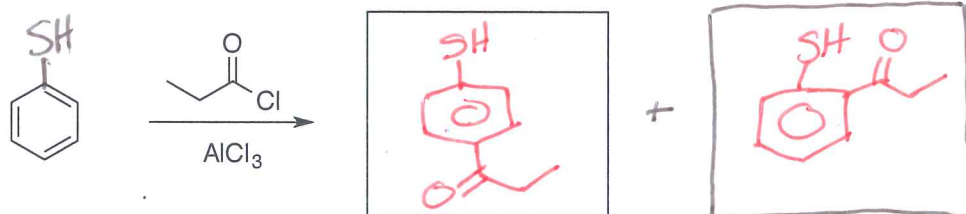
3. (30 points) Single Step Reactions – Choose any five, skip any two reactions ("X" them out). Fill in the missing product or reactants in each reaction.

(a)

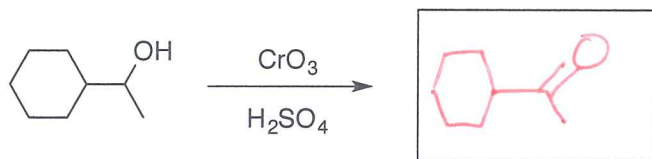


6 pts each

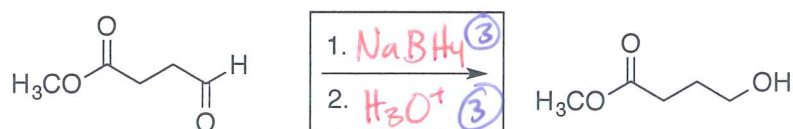
(b)



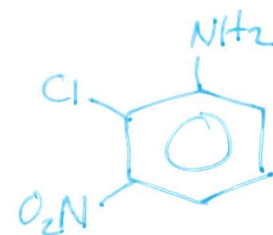
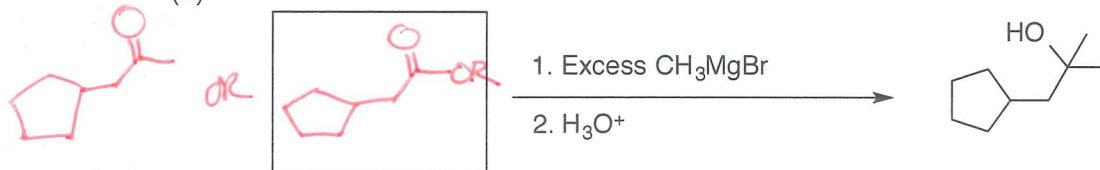
(c)



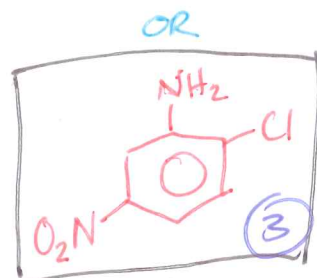
(d)



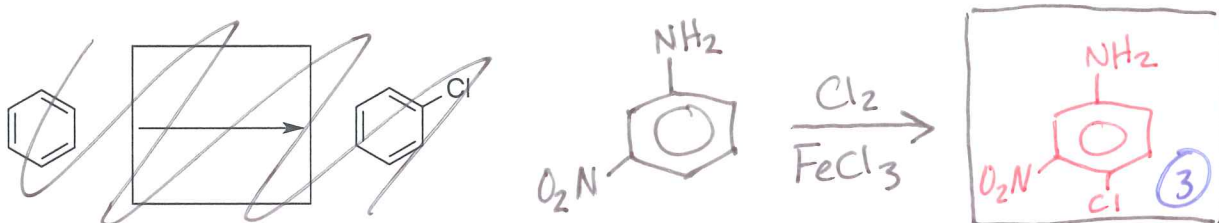
(e)



(f)

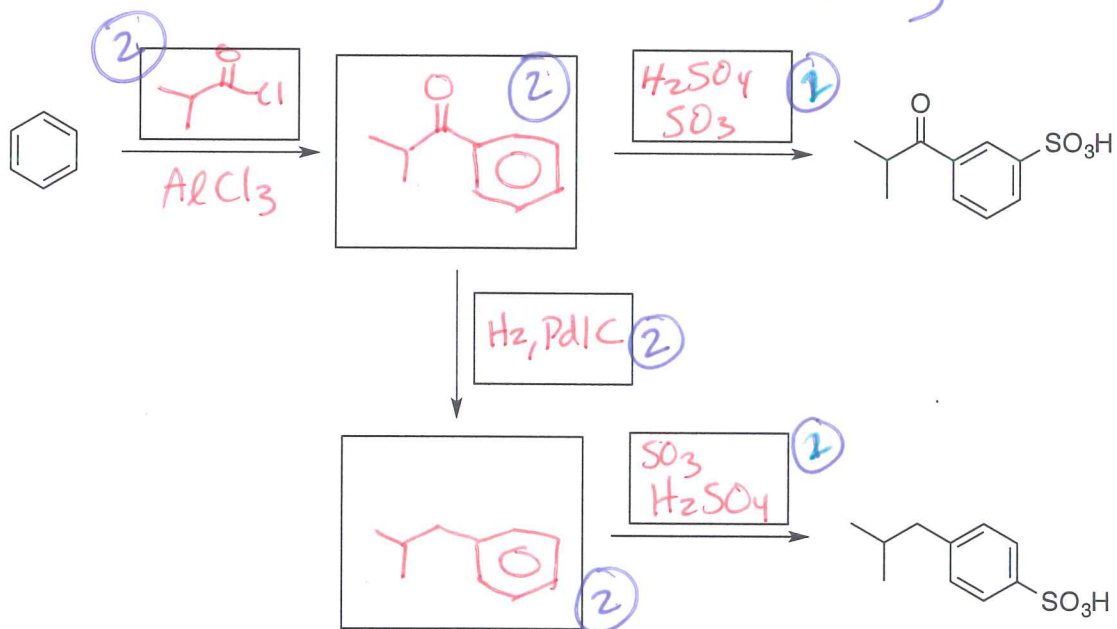
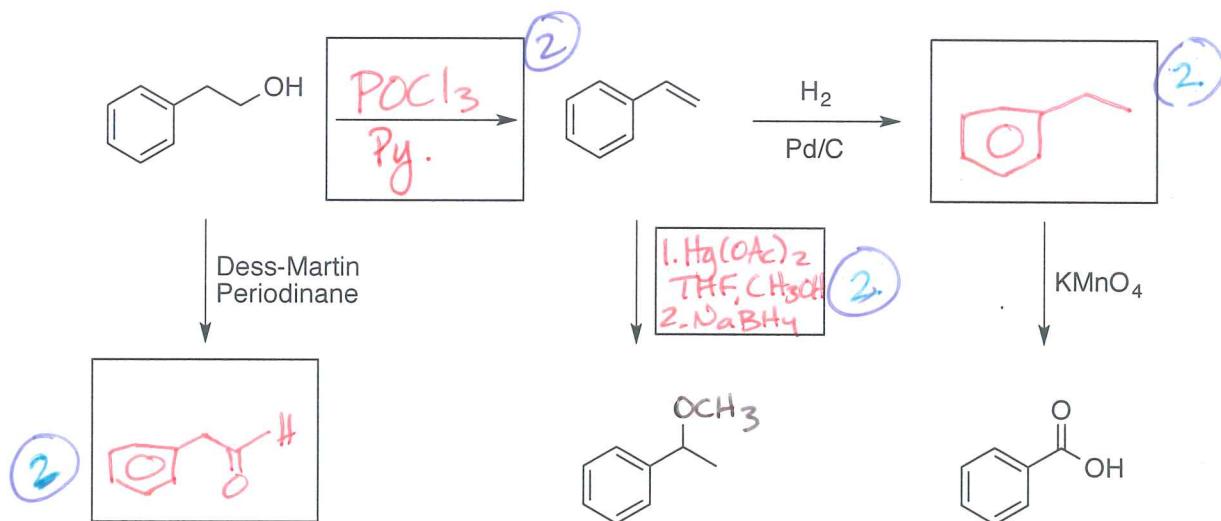


(g)



4. (20 points) Reaction Puzzles – Fill in the missing reagents or product.Puzzle 1

\*has to be FC 1<sup>st</sup> b/c  $\text{-SO}_3\text{H}$  is a strong deactivator

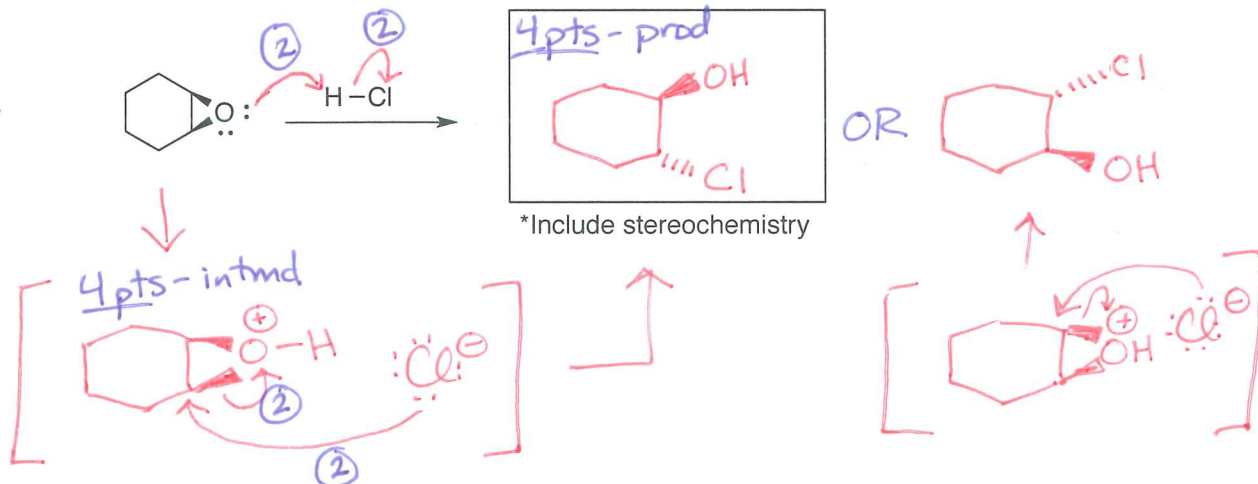
Puzzle 2

5. Mechanisms – complete any two mechanisms. skip one by placing a large X over the entire reaction, otherwise the first two will be graded.

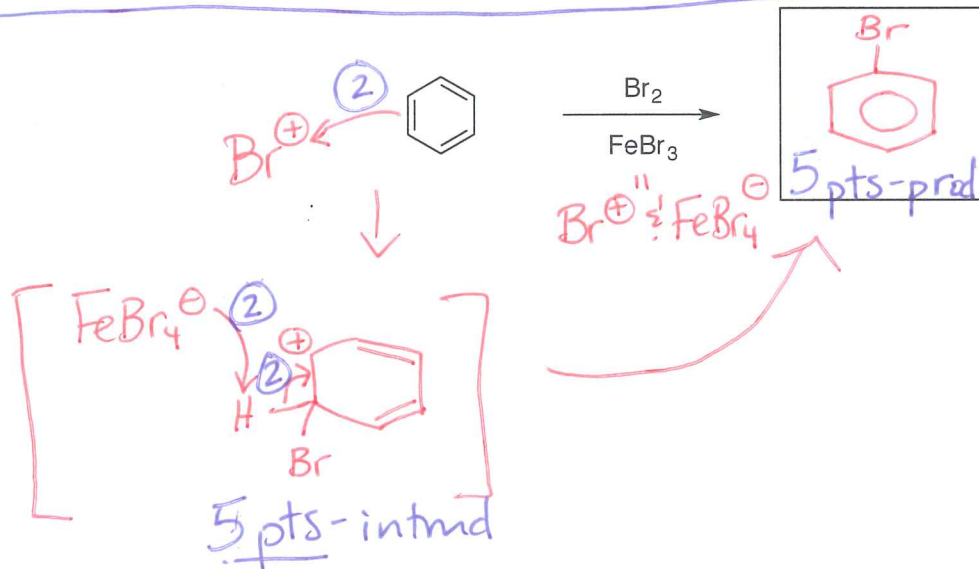
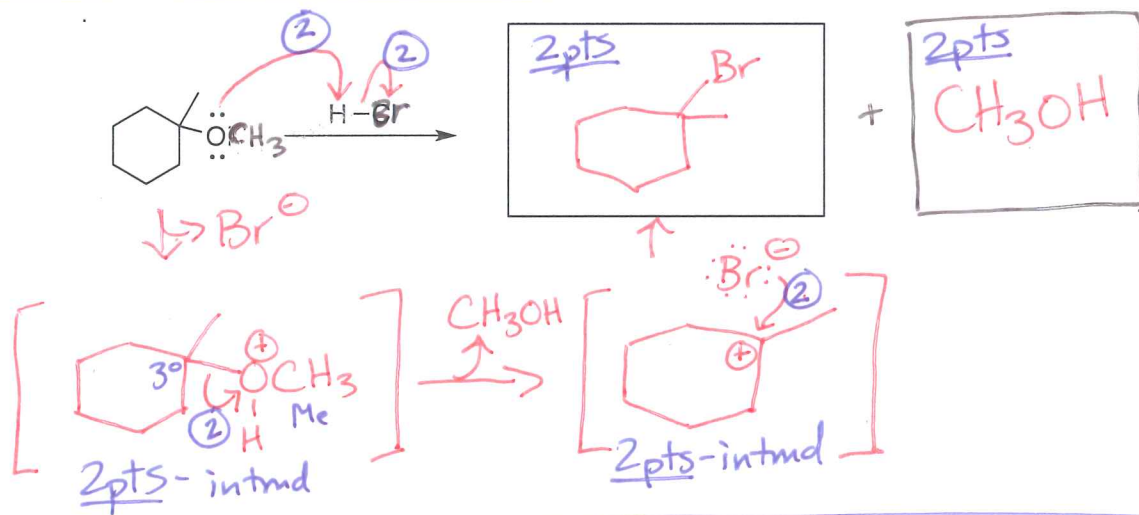
32

(30 points) Show the **product** and full **arrow-pushing mechanisms** for any two reactions (including all acid-base steps, no "PT"). Be sure to clearly indicate all **charged atoms** and **intermediates** after each step.

16 pts each  
possible 2pts extra



\*3°  
S<sub>N</sub>1

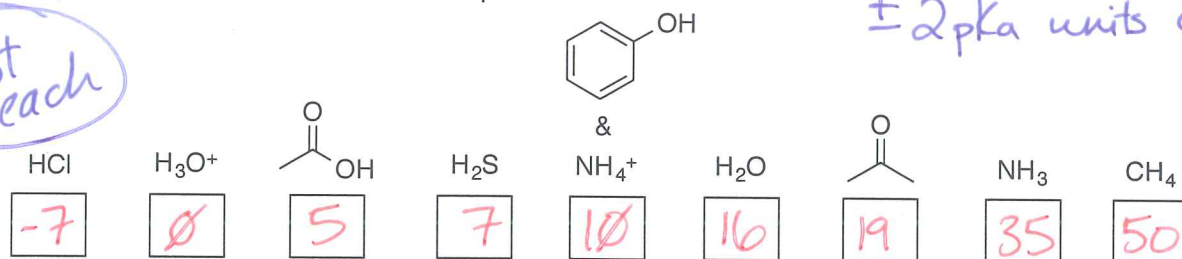


## 2. Acid-Base Chemistry

(a) (9 points) The following compounds are arranged from most (left) to least (right) acidic. Fill in the  $pK_a$  values of each in the boxes provided.

$\pm 2 pK_a$  units OK

1 pt each



(b) (10 points) Rank the following sets of compounds in terms of acidity where 1 is the most acidic. Provide your answer by circling ranking options I, II, or III.

Set 1

5 pts

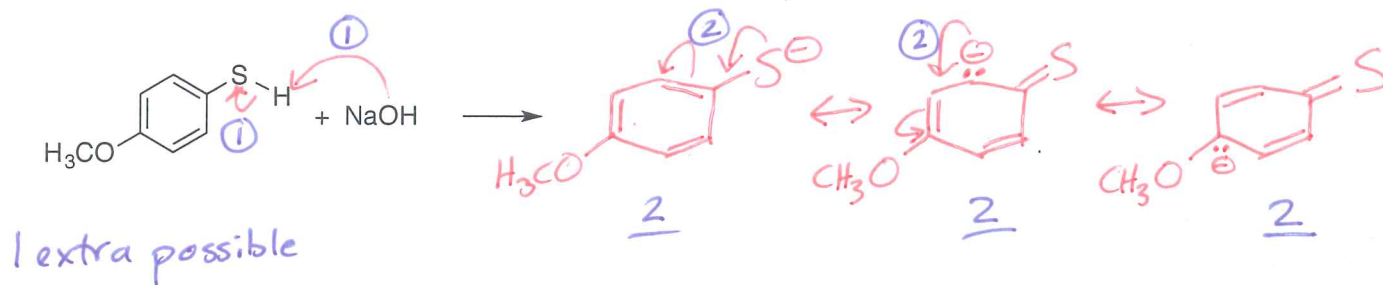
Circle your answer below ↓				
I	2	4	1	3
II	1	3	4	2
III	3	4	1	2

Set 2

5 pts

Circle your answer below ↓				
I	2	3	1	4
II	1	3	2	4
III	2	4	1	3

12  
(c) (11 points) Draw the products of the following reaction and two additional non-equivalent resonance structures of the conjugate base. Include arrow-pushing for each step.



## 6. (50 points) Multi-Step Synthesis – Choose any two

Carry out the synthesis of the indicated target molecules using the starting material provided and any other reagents or sources of carbon needed. **Show the product after each reaction.** No mechanisms. Partial credit is given where possible so if you're stuck, take a deep breath then work your way backwards and/or forwards.

