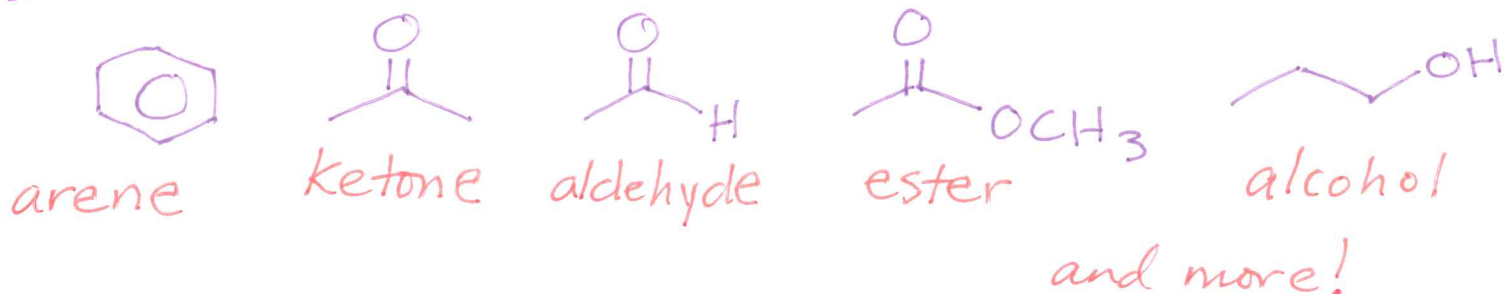


CHEM 8B, Lecture 1

- Mini 8A Refresher
- Chapter 16.1-16.3 : Electrophilic Aromatic Substitution (EArS)

FG's

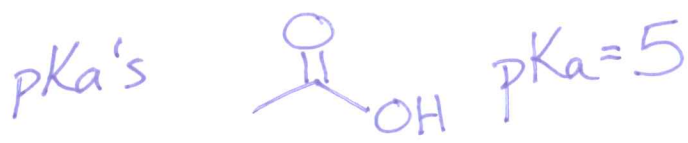
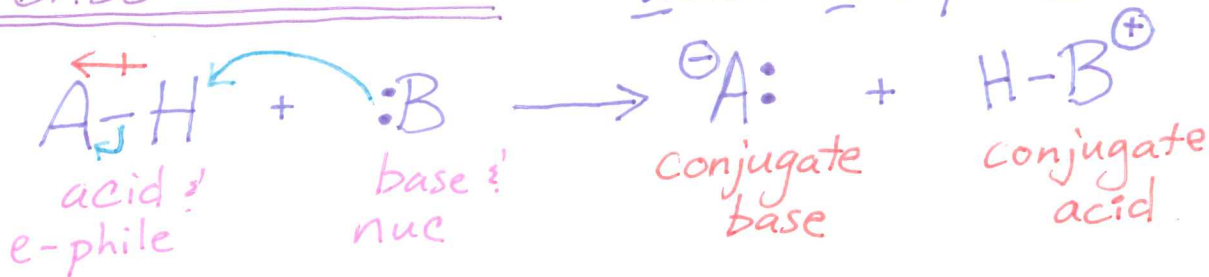


Arrow-Pushing

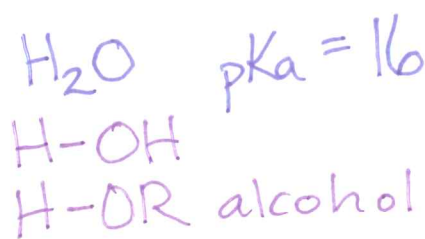
always from e-rich to e-poor
Nuc:  E-phile

ACID-BASE CHEMISTRY

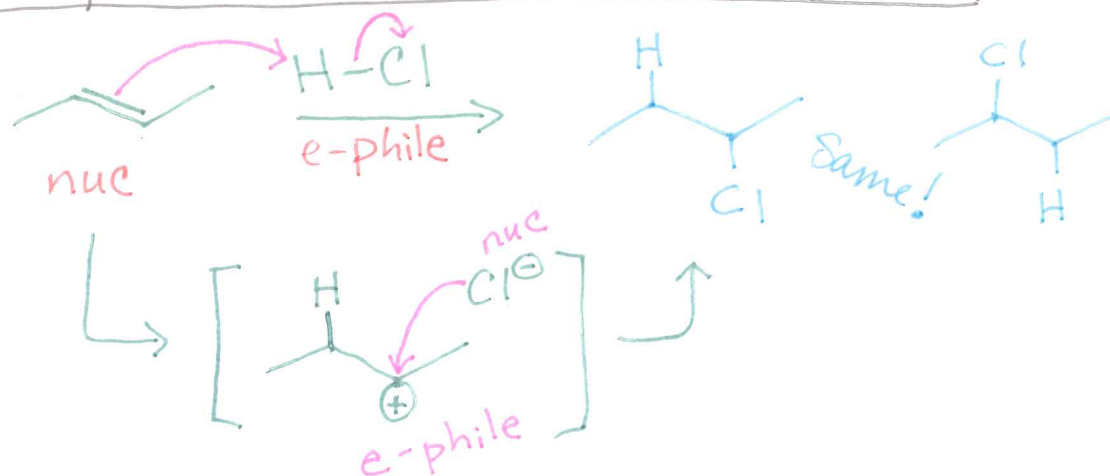
acids accept electrons



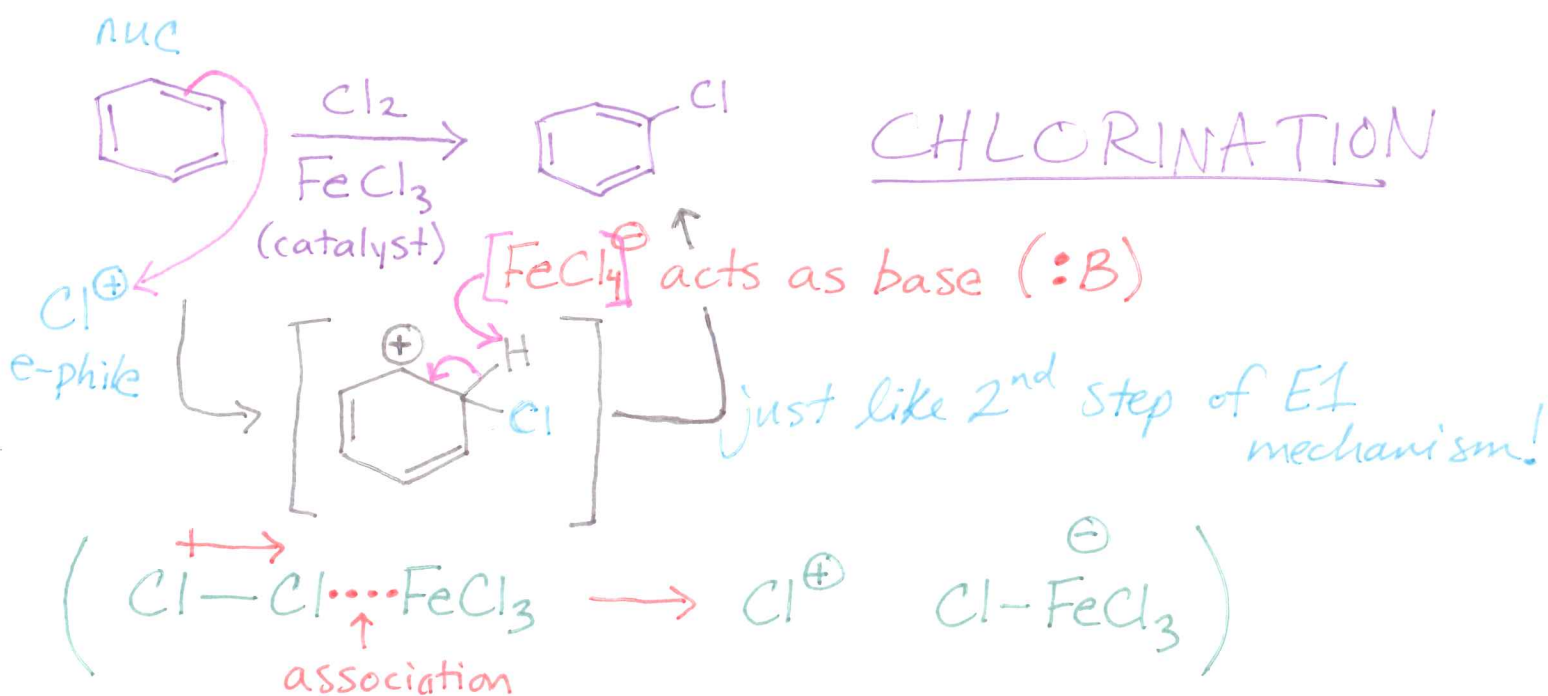
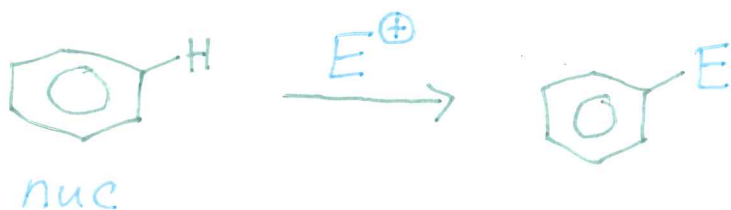
any carb. acid
has pKa ~ 5
and is more acidic than
any alcohol



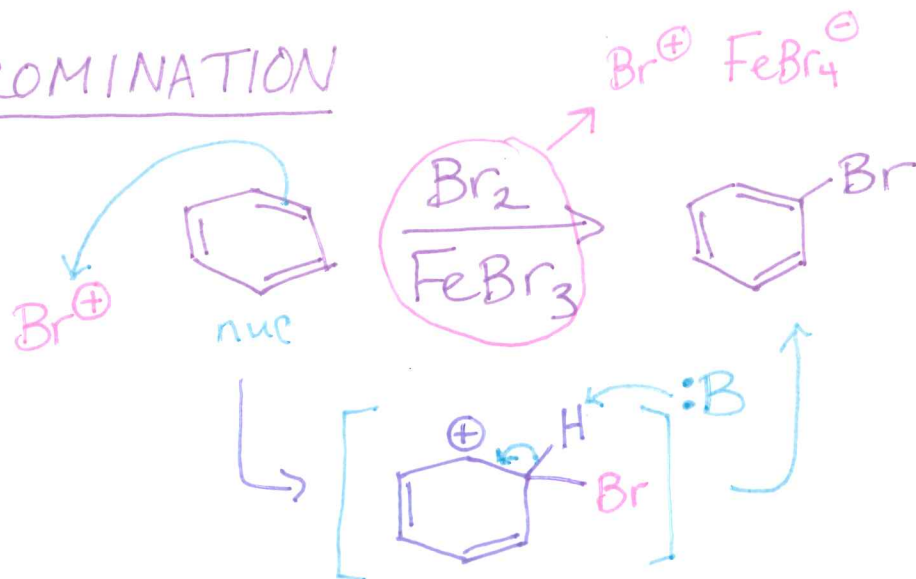
Electrophilic Add'n to Alkenes (Ch 8)



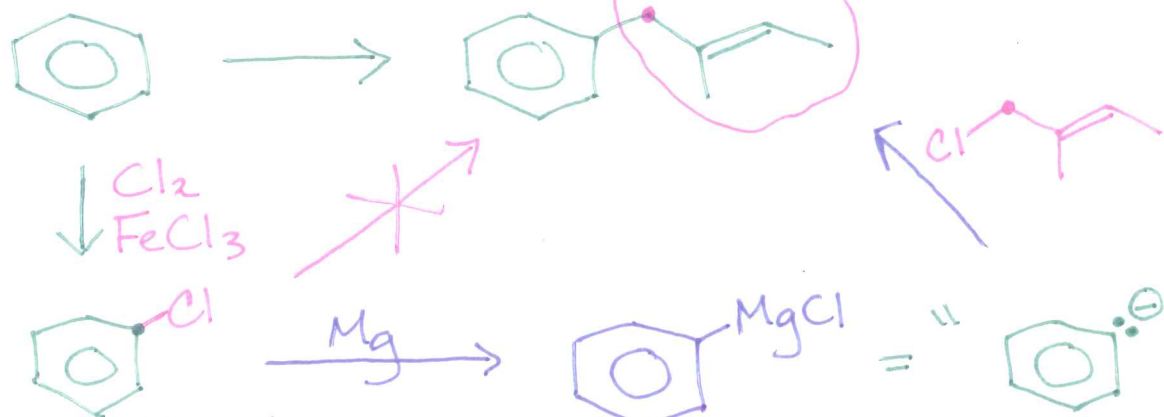
CHAPTER 16 Electrophilic Aromatic Substitution (EArS)



BROMINATION



Why would we do this?
(chemists)

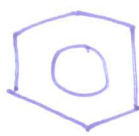
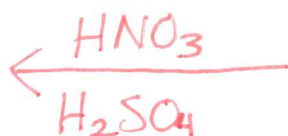
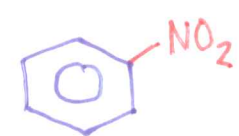


aryl chloride
DO NOT PARTICIPATE
DIRECTLY IN SUB. RXNS

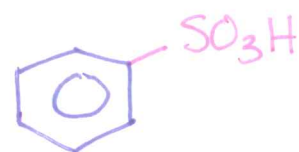
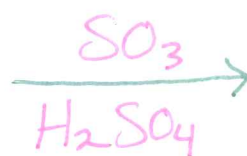
Phenyl magnesium chloride
(GRIGNARD REAGENT)

EArS

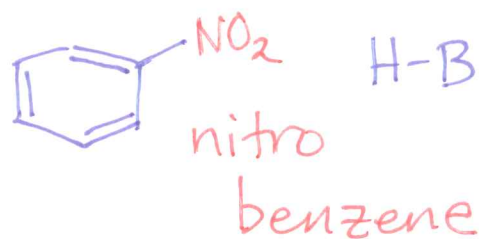
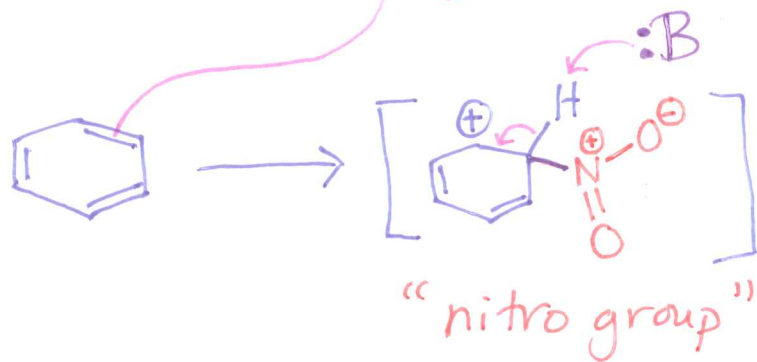
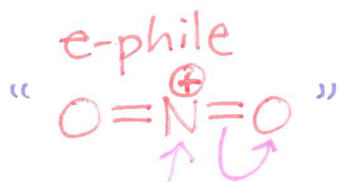
Nitration



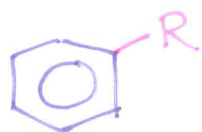
Sulfonation



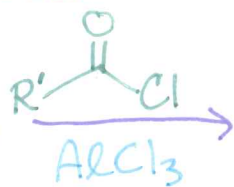
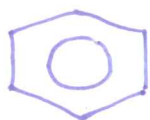
sulfonic acid



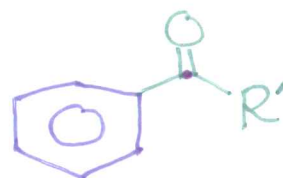
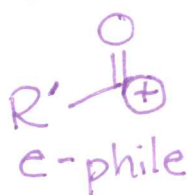
EArS : Friedel-Crafts Alkylation & Acylation *new C-C bonds*



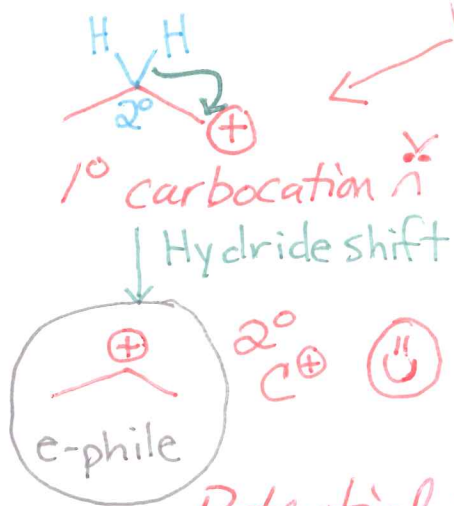
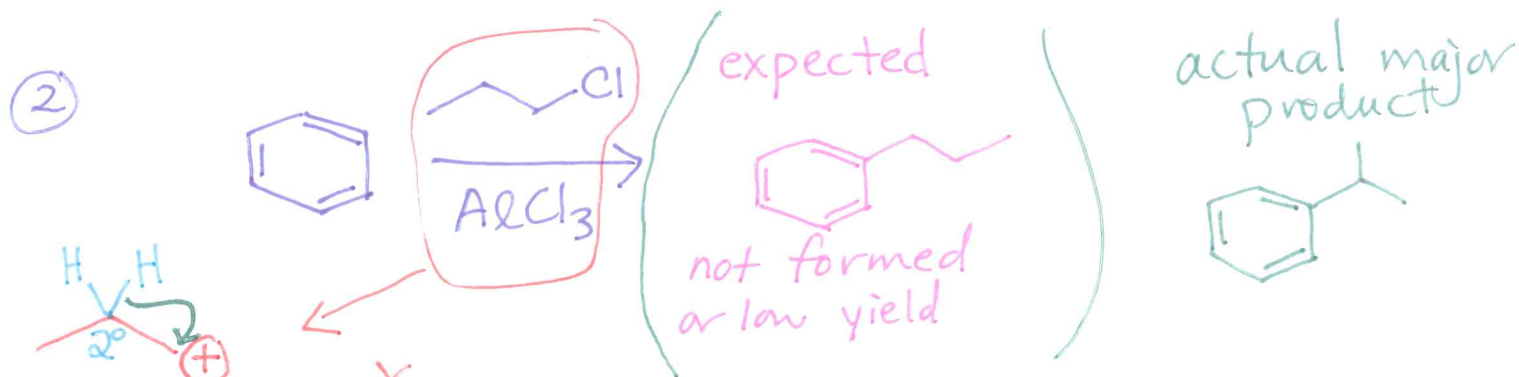
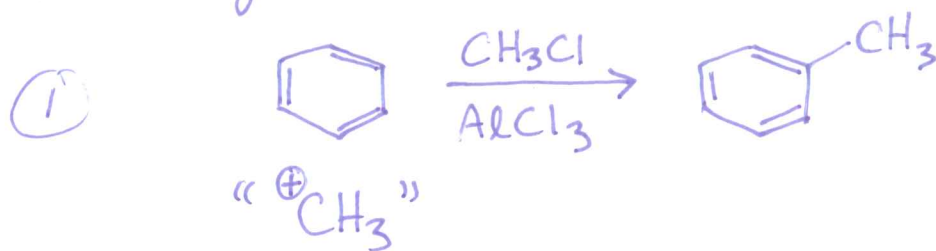
alkylation



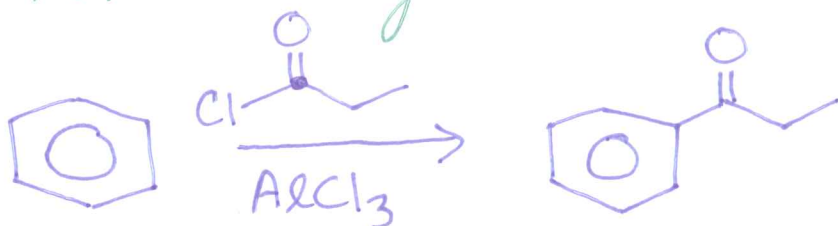
acylation



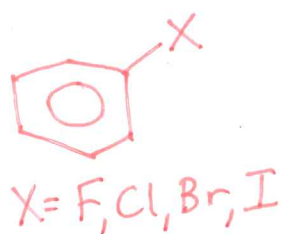
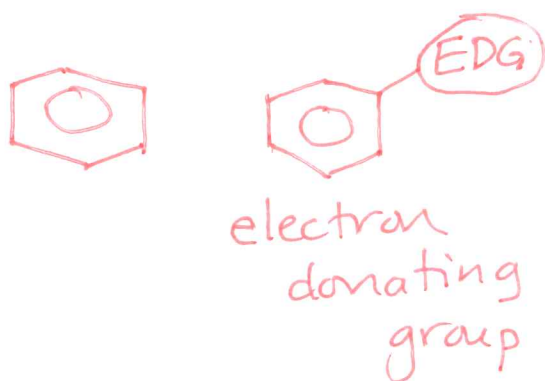
FC Alkylation



Potential for...
...Hydride or alkyl shifts in FC alkylation only,
not FC acylation



All FC rxns require benzene to be "activated"



"deactivated" rings won't react

