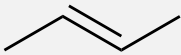
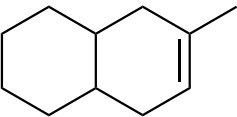
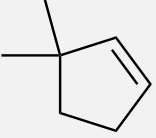
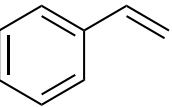

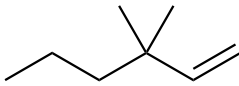

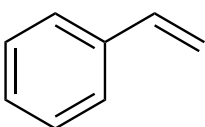
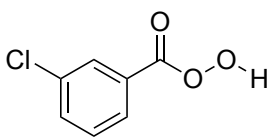


# CHEM 8B Chapters 13-14 Homework – Alkene Reactions and Conjugated/Aromatic Compounds

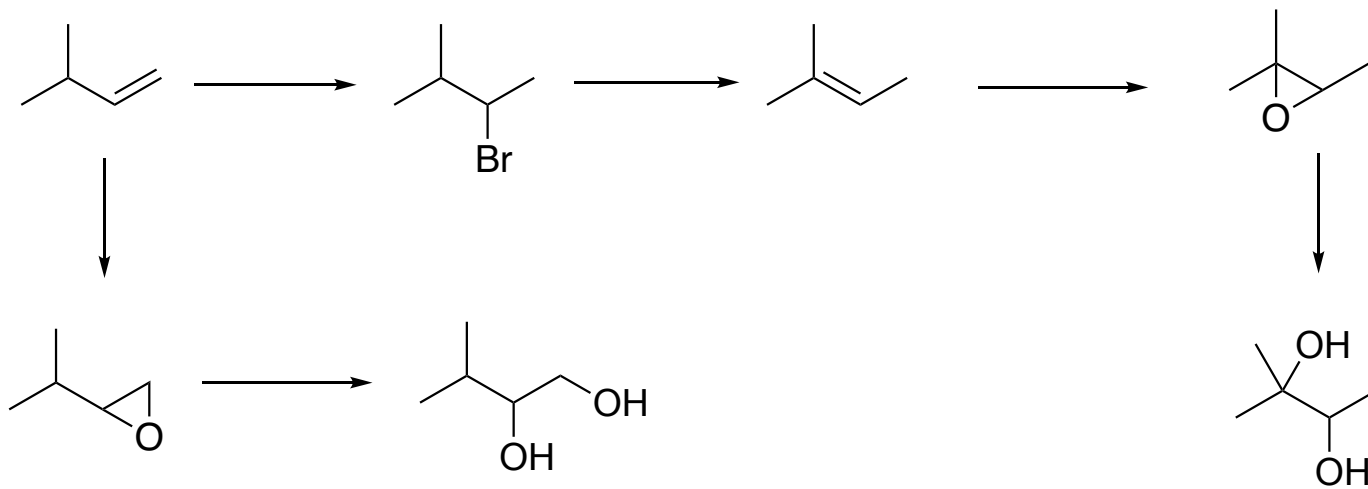
Includes Chapter 12 alkene reactions

- Draw the product of each reaction: **starting material + reagent → Product.**
- Several correct products may be possible. **Draw ONE product to be graded** in the box.

Starting Material	Reagents & translation * know this mechanism	Draw ONE Product	Alternate reagent
1 	* (a) HBr  <i>Hydrobromic acid</i> (Chapter 12 reaction)		Different halide, same mechanism:  <b>HCl</b> , hydrochloric acid  <b>HI</b> , hydroiodic acid
2 	*(b) Cl <sub>2</sub>  <i>Chlorine</i>		Different halide, same mechanism:  Br <sub>2</sub> , bromine  I <sub>2</sub> , iodine
3 	*(c) Cl <sub>2</sub> , H <sub>2</sub> O  <i>Chlorine in water</i>		Different halide, same mechanism:  Br <sub>2</sub> , H <sub>2</sub> O bromine in water  I <sub>2</sub> , H <sub>2</sub> O in water
4 	(d) H <sub>2</sub> , Pd  Hydrogen gas over palladium metal		H <sub>2</sub> with Pt, Ni, Ni <sub>2</sub> B
5 	(e) 1. BH <sub>3</sub> , THF  2. H <sub>2</sub> O <sub>2</sub> , NaOH  <i>Hydroboration with borane in THF, then oxidation with basic peroxide</i>		No alternate reagent; this is a very unique reaction!

	Starting Material	Reagents & translation * know this mechanism	Draw ONE Product	Notes / Alternate reagents
6		(f) 1. $\text{Hg}(\text{OAc})_2$ , $\text{H}_2\text{O}$ 2. $\text{NaBH}_4$ <i>Oxymercuration with mercury (II) acetate, <b>water</b>, then reduction with sodium borohydride</i>		Similar to $\text{H}_2\text{O}$ , $\text{H}_2\text{SO}_4$ - except mechanism does not include carbocation (no hydride or alkyl shift possible)
7		*(g) $\text{H}_2\text{O}$ , $\text{H}_2\text{SO}_4$ <i>Dilute sulfuric acid (Chapter 12 reaction)</i>		$\text{H}_3\text{O}^+$ <b>Hydronium</b> is the reactive species in the mechanism
8		(h) <i>mCPBA</i>  <i>meta-chloroperoxybenzoic acid</i>		peroxy acids peracetic acid $\text{CH}_3\text{CO}_3\text{H}$

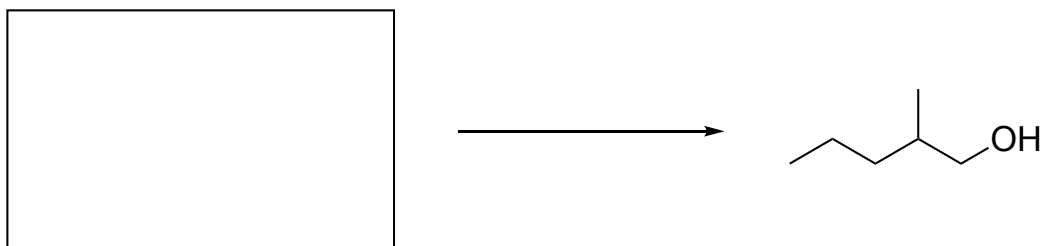
9. Fill in the missing reagent(s) over each arrow.



10. Draw **TWO** potential **alkene(s)** and necessary **reagent(s)** to synthesize this alkyl bromide.

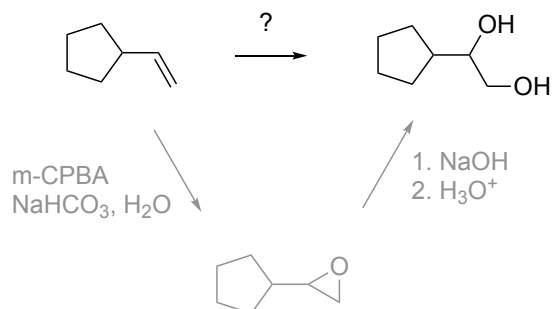


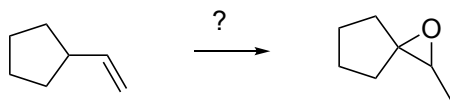
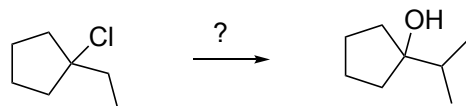
11. Draw the **alkene** and **reagents** needed to synthesize this alcohol.



12. Use your organic reaction toolbelt to propose a **multi-step synthesis** for each transformation below.

- Each problem requires **at least 2 synthetic steps** (reactions).
- **No mechanisms** (curved arrow notation) required.
- Include all **reagents** and draw the **product** of each reaction.
  - Worked Example:

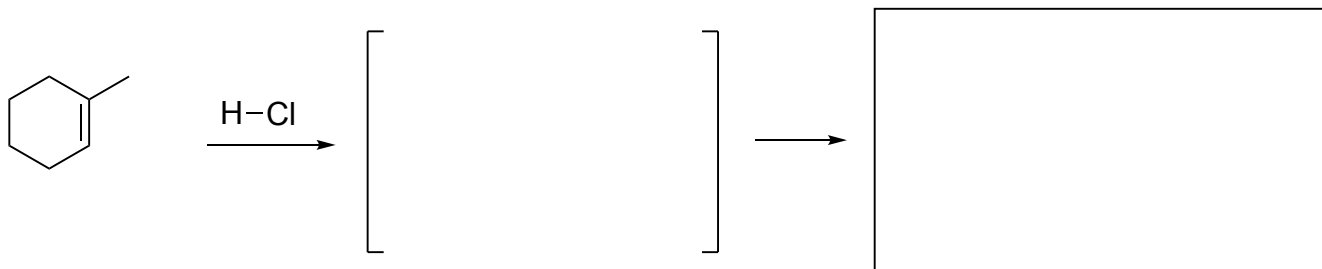


<p><b>(a)</b></p> <div style="text-align: center; margin-bottom: 20px;">  </div> <div style="border: 1px solid black; height: 250px; margin-top: 10px;"></div>	<p><b>(b)</b></p> <div style="text-align: center; margin-bottom: 20px;">  </div> <div style="border: 1px solid black; height: 250px; margin-top: 10px;"></div>
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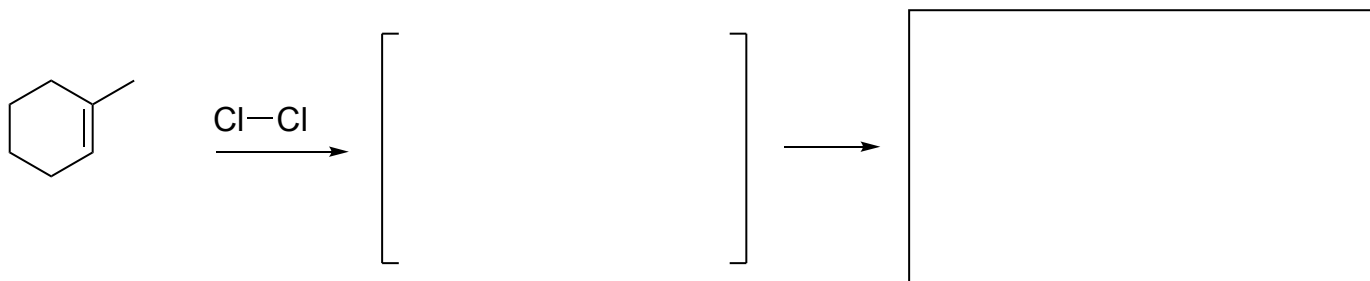
**MORE PRACTICE: Mechanisms**  
*added 1-16-24, NOT required in HW submission*

13. Draw the **mechanisms** for both reactions with curved arrows and intermediate with labeled charges. Draw the **product** of the reaction in the box.

(a)



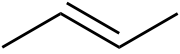
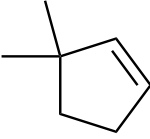
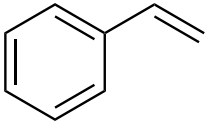
(b)



14. Draw the **mechanism** for this reaction, including curved **arrows** and **intermediates** with labeled charges.



**MORE PRACTICE: Mix & Match with Reaction Bootcamp!**  
*added 1-16-24, NOT required in HW submission*

React each alkene 1-3 with each reagent below and draw the product in the box		1 	2 	3 
(a)	HBr			
(b)	Cl <sub>2</sub>			
(d)	H <sub>2</sub> , Pd			
(e)	1. BH <sub>3</sub> , THF 2. H <sub>2</sub> O <sub>2</sub> , NaOH			
(f)	1. Hg(OAc) <sub>2</sub> , H <sub>2</sub> O 2. NaBH <sub>4</sub>			
(h)	mCPBA			