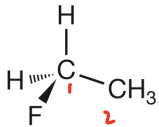
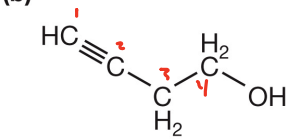
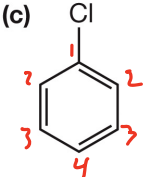
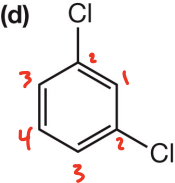
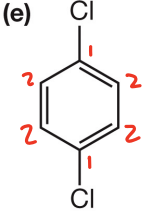
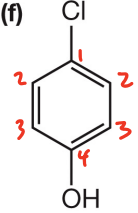


Chapter 17 Homework – Nuclear Magnetic Resonance (NMR)

Part A. Carbon Nuclear Magnetic Resonance, ^{13}C NMR

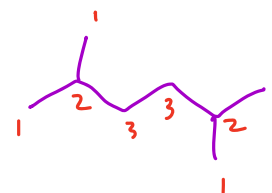
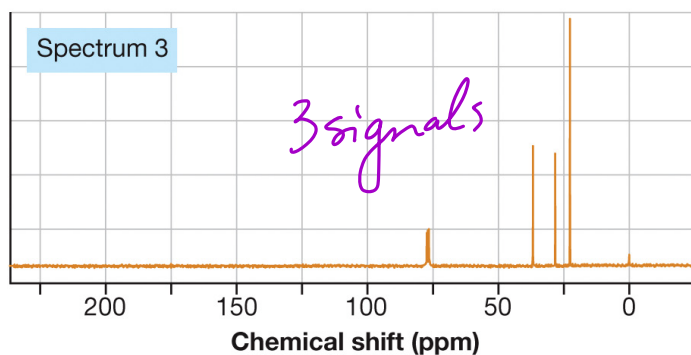
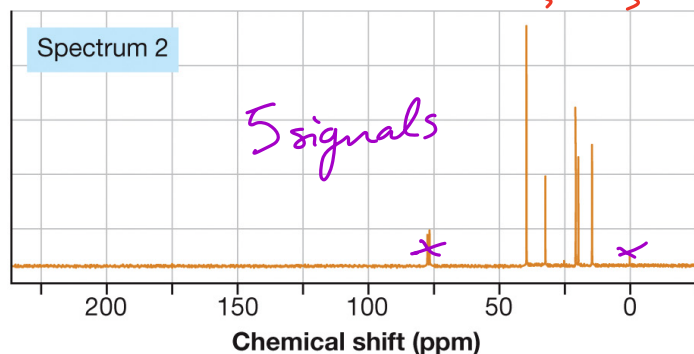
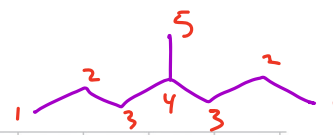
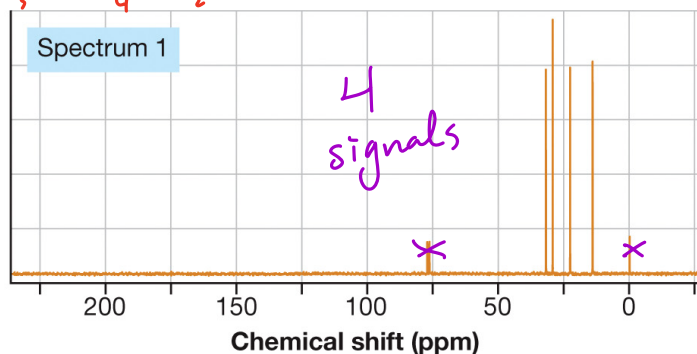
1. How many signals are expected in each compound's ^{13}C NMR spectrum?

(a) (b) (c) (d) (e) (f)	Number of ^{13}C NMR signals
(a) 	2
(b) 	4
(c) 	4
(d) 	4
(e) 	2
(f) 	4

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2. The ^{13}C NMR spectra of three isomers with molecular formula C_8H_{18} are given below. Match each chemical name to its spectrum (1-3). Ignore the peak at 76 ppm (solvent).

Isomer	Spectrum (1, 2, or 3?)
2,5-dimethylhexane	3
4-methylheptane	2
Octane	1



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Ch 17A. ^{13}C NMR

3. How many signals are expected in each compound's ^{13}C NMR spectrum?

A	B	C	D
4	6	6	6

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4. Indicate the **approximate chemical shift (range)** of each highlighted carbon in its ^{13}C NMR spectrum (Table 17-5).

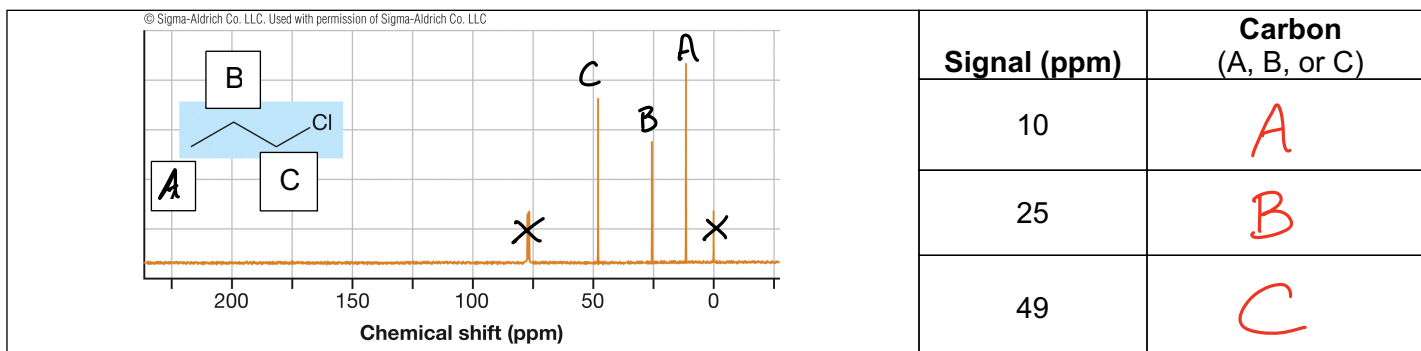
(a) (b) (c)

$\underline{110-160}$ $\underline{\text{CH}}$ $\underline{\text{CH}_3}$ $\underline{20-35}$
 CH CH_2

5. Indicate the expected ^{13}C NMR chemical shift range for each carbon (A-E) in the table below.

Signal	Chemical shift range (ppm)	
	Karty Table 17-5 / Mohrig NMR	
A	10-25	7-30
B	20-45	15-40
C	55-70	55-80
D	160-185	160-180
E	128	110-160

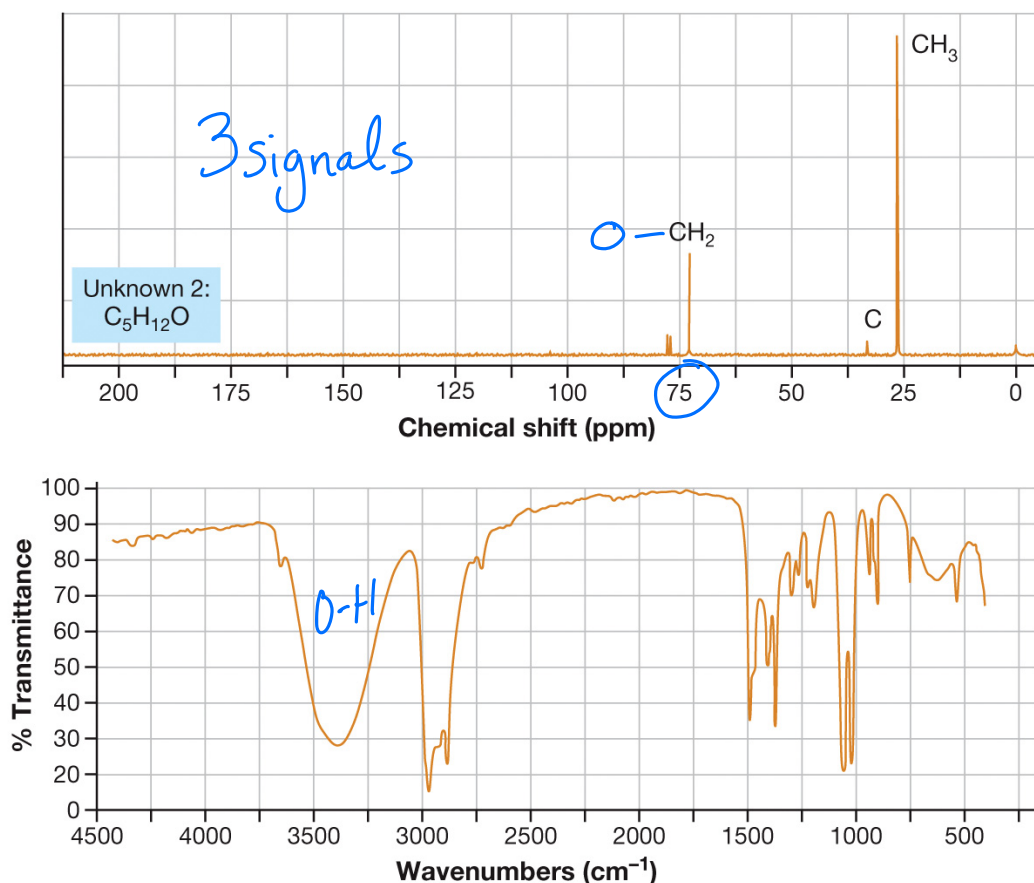
6. 1-chloropropane produced the ^{13}C NMR spectrum shown here. Match each carbon in the molecule (letters A-C) to each signal in the spectrum.



Ch 17, Part A. ^{13}C NMR

7. Propose the structure of the molecule that matches the spectral data below.

see class notes



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What clues does the ^{13}C NMR spectrum provide about the structure?

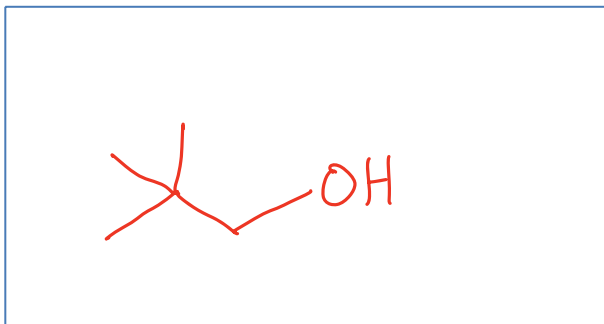
5 C's, 3 signals $\rightarrow$ symmetry

O-CH₂
C (no H's)
CH₃

What clues does the IR spectrum provide about the structure?

alcohol O-H

Structure:



Chapter 17 HW, Part B. Proton Nuclear Magnetic Resonance, ^1H NMR

8. How many signals are expected in each compound's ^1H NMR spectrum?

				Number of ^1H NMR signals			
(a)		(b)		(c)		(d)	
(e)		(f)		(a)	2		
				(b)	4		
				(c)	3		
				(d)	3		
				(e)	1		
				(f)	3		

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9. Indicate the approximate ^1H NMR chemical shift for each carbon (A-E) in the table below.

Signal	Chemical shift (ppm)	
	Karty Table 17-1	Mohrig NMR
A	0.9	0.8-1.9
B	1.3	0.8-1.9
C	3.3	3.2-5.3
D	7.3	6.9-9
E	7.3	6.9-9

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10. How many signals are expected in each compound's ^1H NMR spectrum?

A	B	C	D
2	3	3	3

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Chapter 17B. ^1H NMR

11. Rank protons A-C in order from **largest to smallest chemical shift**. Write the proton's letter in the right column of the table. Then, provide the **integration value** (number of H's) responsible for each signal.

Chemical Shift Rank	Proton (A-C)	Integration (# of H's)
High	A	1H
Medium	C	2H
Low	B	2H

12. Rank protons D-H in order from **largest to smallest chemical shift**. Write the proton's letter in the right column of the table.

Chemical Shift Rank	Proton (D-H)	Integration (# of H's)
Highest	D	1
	G	1
	E	2
	F	2
Lowest	H	3

13. Rank protons I-M in order from **largest chemical shift to smallest**. Write the proton's letter in the right column of the table.

Chemical Shift Rank	Proton (I-M)	Integration (# of H's)
Highest	L	1H
	K	1H
	M	2H
	J	2H
Lowest	I	3H

14. Report the approximate ^1H NMR and ^{13}C NMR chemical shifts of just the CH_3 group in similar benzylic esters **E** and **F** below.

	^1H NMR (Table 17-1)	^{13}C NMR (Table 17-5)
E	3.3	55-70
F	2.1	30-45

Chapter 17B. ^1H NMR

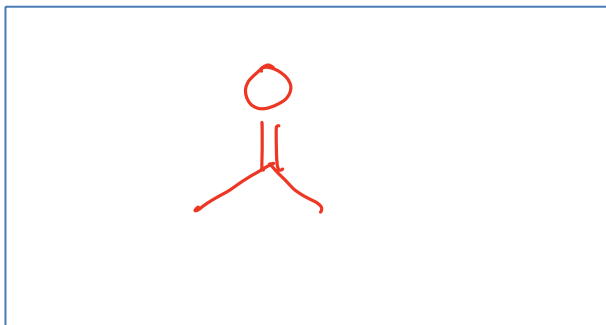
15. Use the $N+1$ rule, where N is the number of adjacent H's, to predict **splitting pattern** of the highlighted signal in its ^1H NMR spectrum (singlet, doublet, triplet, quartet, pentet, sextet, septet, octet, or nonet).

		$N+1$ Splitting Pattern(s)
(a)		1, singlet
(b)		9, nonet
(c)		CH ₃ group: 3, triplet CH ₂ group: 4, quartet
(d)		2, doublet
(e)		CH ₂ group: 3, triplet CH ₃ group: 1, singlet
(f)		3, triplet

16. **Structural Elucidation:** use the data below to determine the structure of the compound.

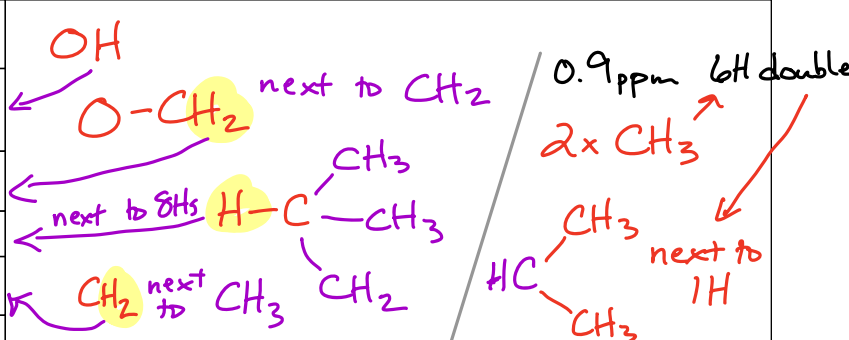
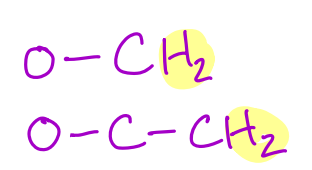
		What does each piece of data tell you about the compound?
Formula	$\text{C}_3\text{H}_6\text{O}$	Any double bonds? yes!
IR	1720 cm^{-1} (broad, strong) 2900 cm^{-1} (sharp, strong)	C=O C-H alkane
^1H NMR	One signal... Chemical shift = 2.0 ppm Integration = 6H Splitting = singlet	2 x CH₃ O=C-CH₃
^{13}C NMR	207 ppm 31 ppm	C=O C-C=O

Structure:

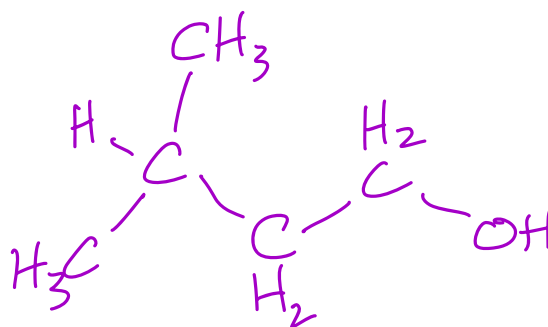
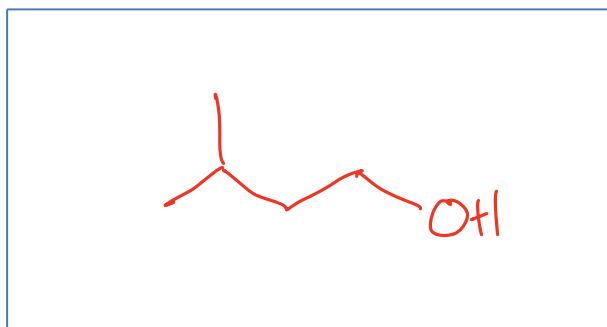


Chapter 17 HW

17. Structural Elucidation: use the data below to determine the structure of the compound.

Formula	$C_5H_{12}O$ <i>no double bonds</i> 			What does each signal tell you about the compound? (show your work in the space below)
IR	3300 cm^{-1} 2900 cm^{-1}	<i>OH</i> <i>C-H alkane</i>		
1H NMR	Chemical shift	Integration (# of Hs)	Splitting	
	4.0 ppm	1 <i>CH</i>	Broad singlet	
	3.5 ppm	2 <i>CH₂</i>	Triplet	
	1.6 ppm	1 <i>CH</i>	nonet	
	1.5 ppm	2 <i>CH₂</i>	quartet	
	0.9 ppm	6 <i>2x CH₃</i>	doublet	
^{13}C NMR	? 61 (CH_2) <i>close to O</i> ? 42 (CH_2) <i>kind close to O</i> ? 25 (CH) <i>not close to O</i> ? 23 (CH_3) <i>to O</i>			

Structure:



This graded HW set includes problems adapted from the Karty 3 text.

Recommended Problems from Karty 3, Chapter 17 – see Canvas textbook module

- **In-Chapter Problems**, “Your Turn” = 1-4, 7, 8, 14-16, 20, 21, 24-27, 30
- **End-o-Chapter Problems** = 1-2, 6-9, 11, 12, 21, 24, 25, 29-35, 41, 44, 47, 49, 52, 53
- Solutions to all problems are now available for FREE in the Canvas textbook module!