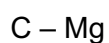


Chapter 18-19 Homework – Addition to Polar pi Bonds

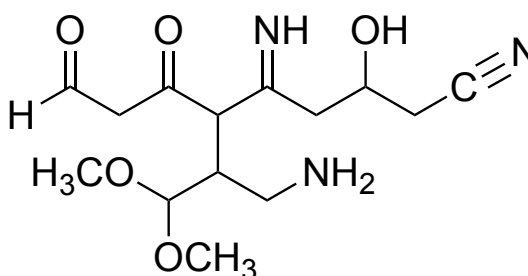
Chapter 18A. Bond Basics

1. to each bond to indicate its polarity.

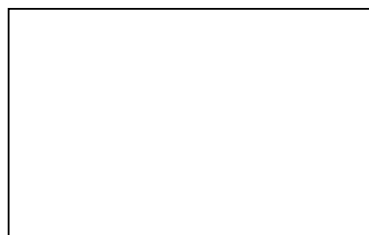
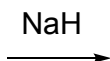
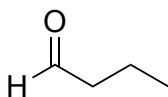


2. Circle and label each functional group in the fictional molecule below.

- Acetal
- Alcohol
- Aldehyde
- Amine
- Imine
- Ketone
- Nitrile

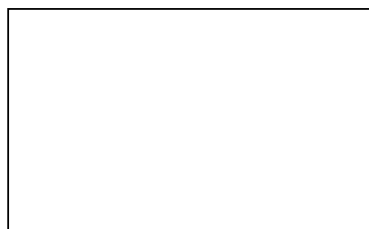


3. Show the mechanism and product for the **alpha-deprotonation of butanal**.
- **Draw the H's** in the alpha position,
 - use **curved arrow** notation to show the proton transfer reaction with sodium **hydride**,
 - and **draw the enolate** formed.

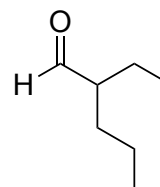
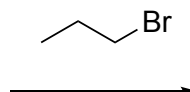


enolate

4. React the enolate above with propylbromide via S_N2 mechanism (one-step substitution).
- **Redraw the enolate** from #3 above.
 - **Add curved arrows** to explain how the bonds are broken and formed.

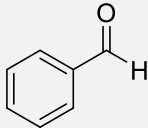
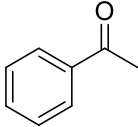
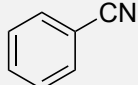
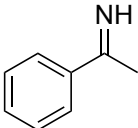
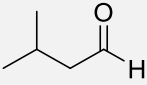
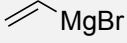

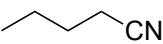
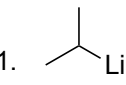
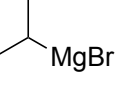


enolate



18B. CARBONYL REACTIONS

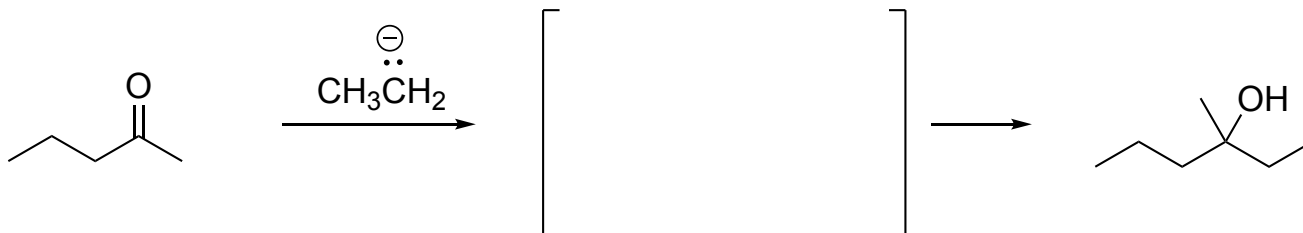
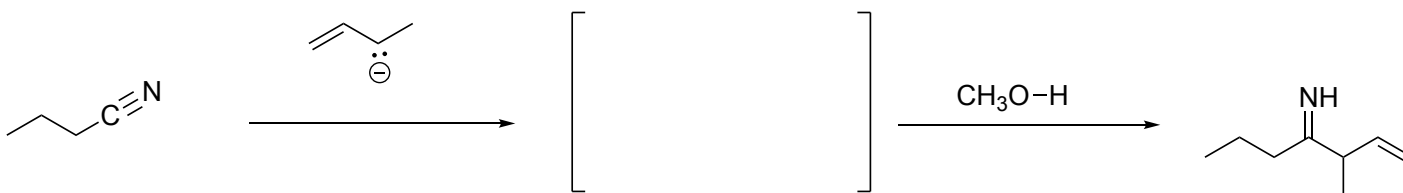
- Draw the product of each reaction: **starting material + reagents → Product.**

	Starting Material	Reagents & translation *be able to draw the arrow-pushing mechanism	Alternate reagents (same product)	Draw the Product
5	 benzaldehyde (almond extract)	NaBH₄, MeOH <i>sodium borohydride in methanol</i>	1. NaBH ₄ 2. H ₂ O <i>Or</i> 1. LiAlH ₄ 2. H ₂ O	
6	 Acetophenone	1. LiAlH₄ 2. H₂O <i>lithium aluminum hydride followed by water</i>	NaBH ₄ , MeOH <i>Or</i> 1. NaBH ₄ 2. H ₂ O	
7	 benzonitrile	1. LiAlH₄ 2. H₂O <i>lithium aluminum hydride followed by water</i>	n/a	
8	 imine	NaBH₄ <i>sodium borohydride in methanol</i>	1. LiAlH ₄ 2. H ₂ O	
9	 3-methylbutanal	1.  MgBr 2. H ₂ O <i>vinyl magnesium bromide followed by water</i>	1.  Li 2. H ₂ O	
10	 nitrile	1.  Li 2. H ₂ O <i>Isopropyl lithium followed by water</i>	1.  MgBr 2. H ₂ O	


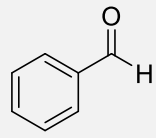
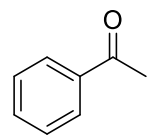
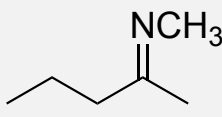
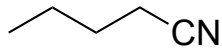
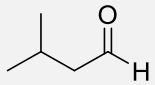
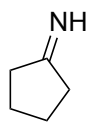
Pro-tip: See the REACTION SUMMARY at the end of Chapter 18 class notes.

Ch 18C. POLAR π BOND ADDITION MECHANISMS

- Draw the arrow-pushing mechanism for the reactions, including all charged intermediates and product.
- Hydride and organometallic reagents are simplified with their nucleophilic form.

11. Ketone Reduction**12. Nitrile reduction****13. Imine reduction****14. Addition of organometallic to aldehyde/ketone****15. Nitrile + organometallic**

18D. Mix & Match with Reaction Bootcamp!

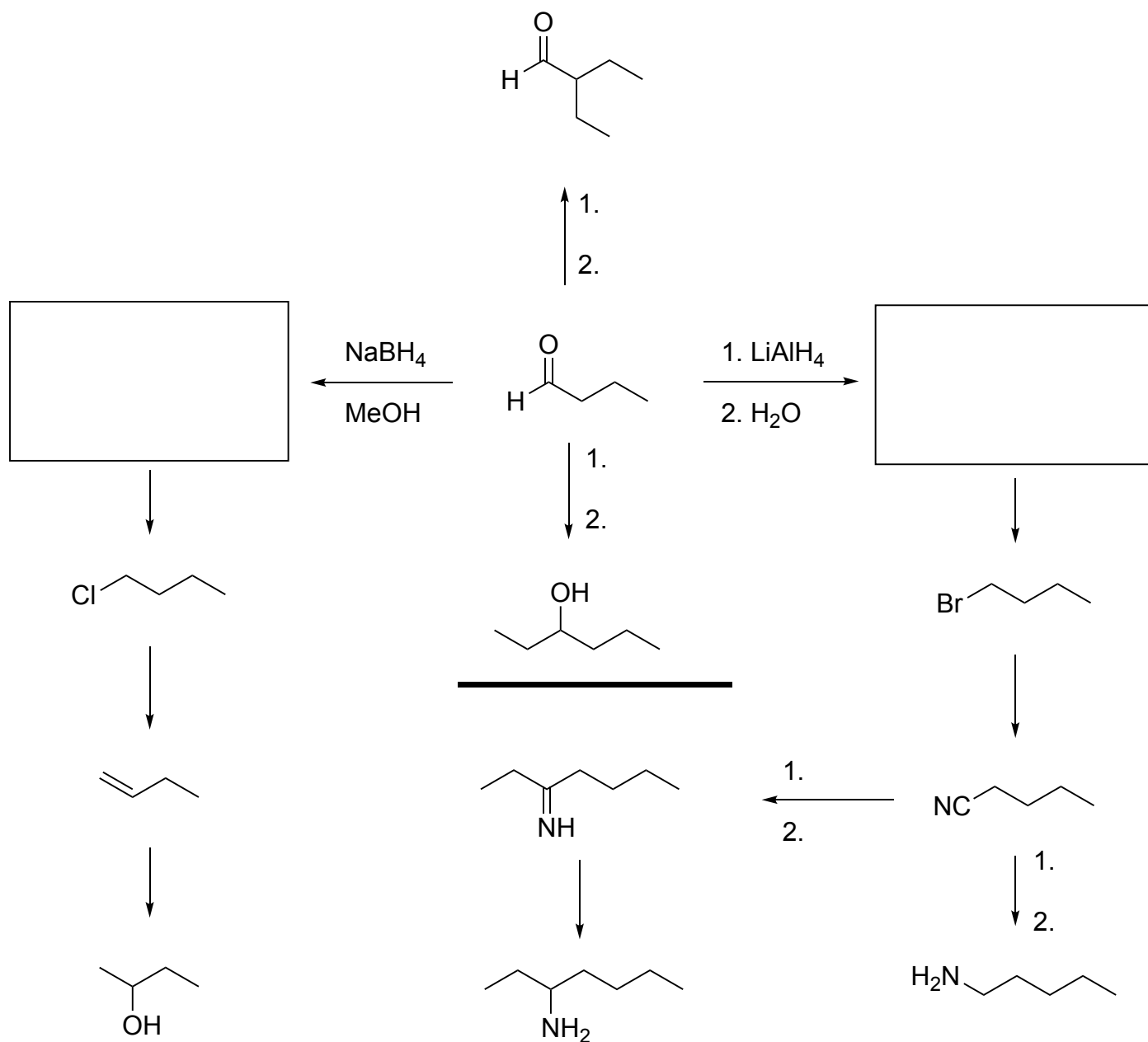
	React each carbonyl compound with each reagent and draw the product in the box	NaBH ₄ , MeOH	1. LiAlH ₄ 2. H ₂ O	1.  MgBr 2. H ₂ O
16	 benzaldehyde (almond extract)			
17	 Acetophenone			
18	 imine			NO REACTION
19	 pentanenitrile	NO REACTION		
20	 3-methylbutanal			
21	 imine			NO REACTION

18E. Reaction Puzzle - “training wheels” for multi-step synthesis

The “puzzle” below covers Chapter 18 and previous reactions. Take it one step at a time.

Draw the missing products in the boxes and **add missing reagents** to the arrows.

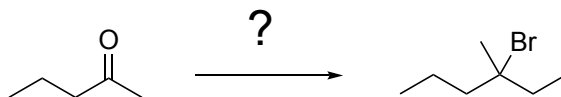
When the reagents need to be separated into steps, the numbers are provided for you.



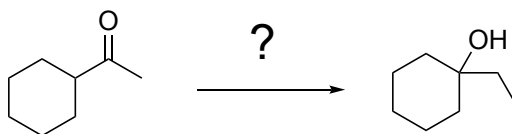
18F. Multi-Step Synthesis

- Each transformation requires **at least two synthetic steps** to reach the target product.
 - These problems were designed to use no more than four reactions.
 - There are multiple pathways and it's ok if you use a feasible pathway with more than four steps ☺
- Show each set of **reagents and reaction products** on the journey.
 - Mechanisms are not required, but may be helpful.
- If there is a mixture of products (ex. *major* and *minor*), assume the minor product can be removed.
 - You can just draw the desired *major* product.

22.



23.



24.

