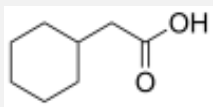
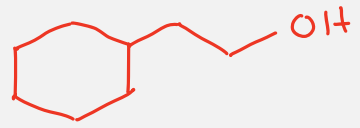
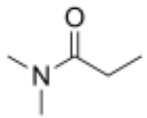
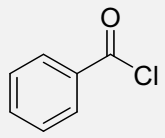
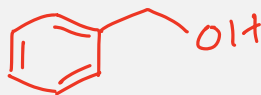
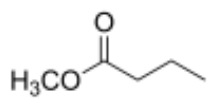
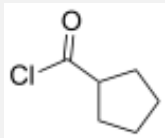
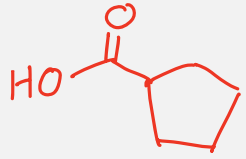
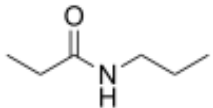
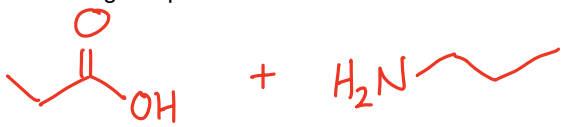
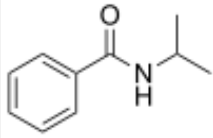


Chapter 22 Homework – Carboxylic Acids & Friends

20A. HYDROGEN & OXYGEN NUCLEOPHILES.

Draw the product of each reaction: starting material + reagent → Product.

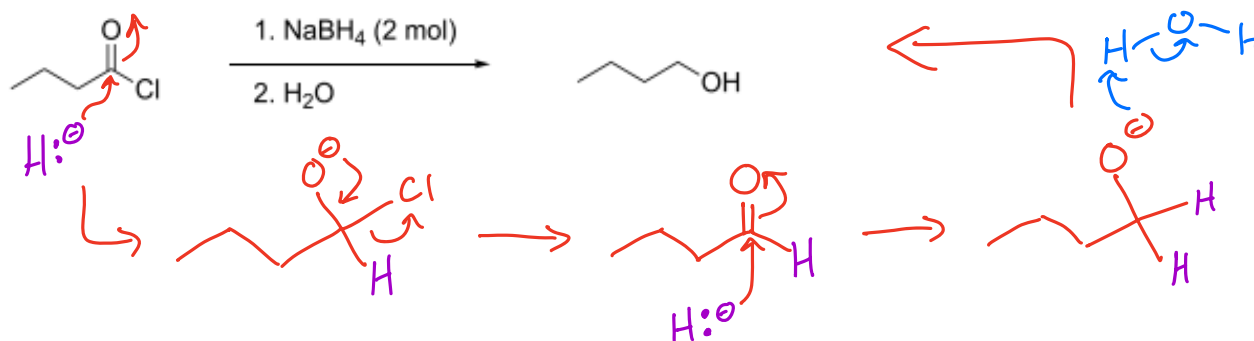
	Starting Material	Reagents & translation	Draw the Product
1		1. xs LiAlH ₄ 2. H ₂ O <i>Excess lithium aluminum hydride, followed by water</i>	
2		1. xs LiAlH ₄ 2. H ₂ O	
3		xs NaBH ₄ , MeOH <i>sodium borohydride in methanol</i>	
4		1. xs LiAlH ₄ 2. H ₂ O	Draw both organic products 
5		H ₃ O ⁺ , <i>Aqueous acid and heat</i>	
6		H ₃ O ⁺ ,	Draw both organic products  <i>acidic conditions, protonated amine</i> 
7		H ₃ O ⁺ ,	Draw both organic products 



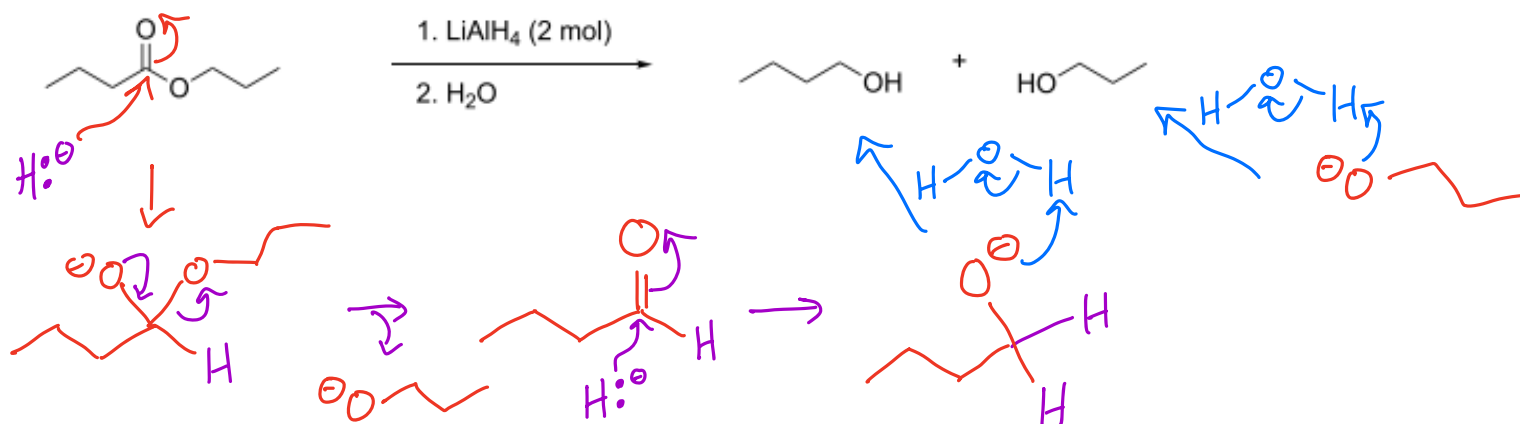
22B. Mechanisms – Acid Derivatives with hydrogen and oxygen nucleophiles.

- Draw the arrow-pushing mechanism for each reaction, including all charged intermediates and product.

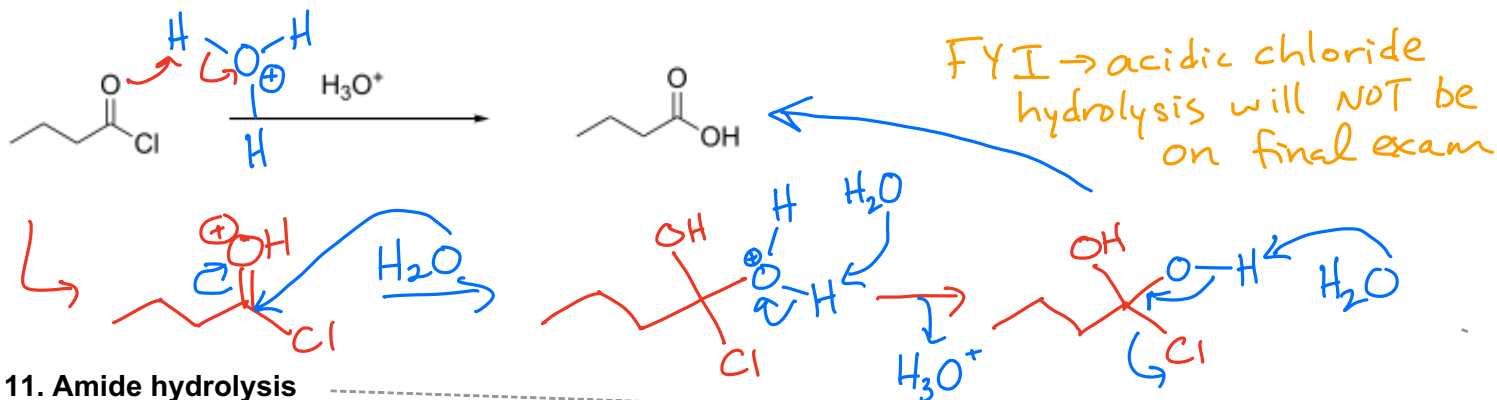
8. Acid chloride reduction with excess LAH



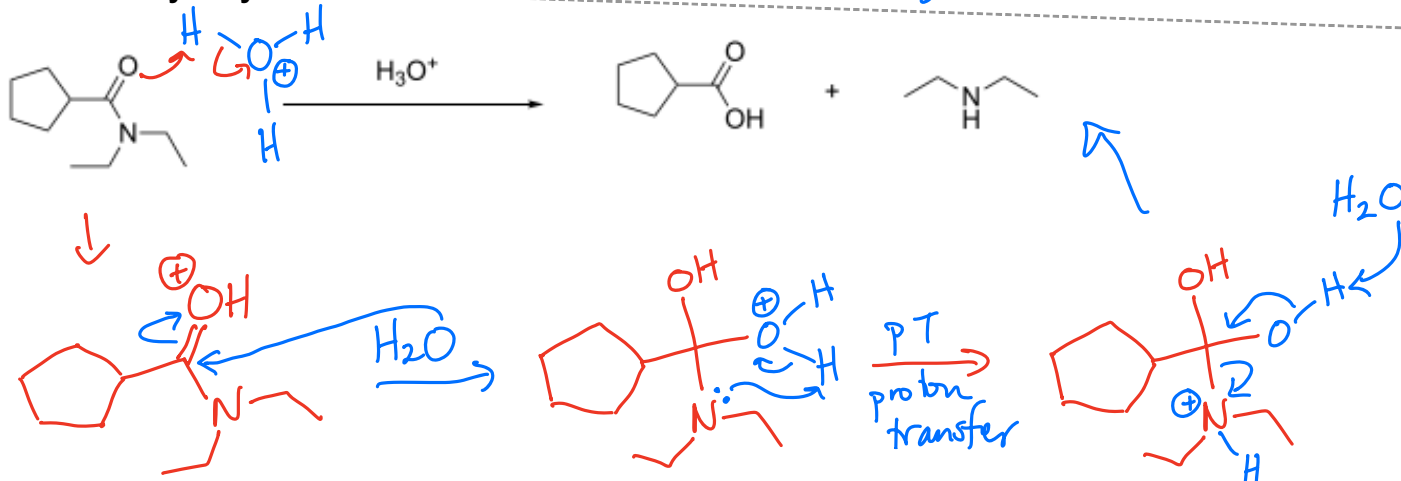
9. Ester reduction with excess LAH



10. Acid chloride hydrolysis



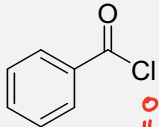
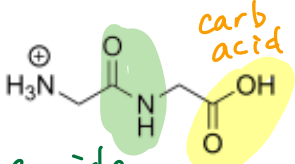
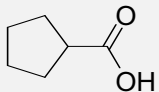
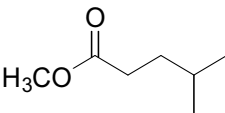
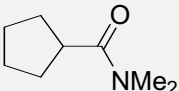
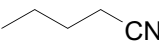
11. Amide hydrolysis



22C. Nucleophilic Acyl Substitution – Mix & Match with Reaction Bootcamp!

☐ Draw the product(s) of each reaction: **starting material + reagent** ☐ **Product(s)**

☐ **Look out for “No Reaction”** – when the reagent does not react with the starting material

React each molecule with each reagent and draw the product in the box		mild reduction xs NaBH_4 , MeOH	hydrolysis H_3O^+	strong reduction 1. xs LiAlH_4 2. H_2O
12	 <chem>Ph-COCl</chem>	<chem>Ph-CH2OH</chem>	<chem>Ph-COOH</chem>	<chem>Ph-CH2OH</chem>
13	 amide dipeptide carb acid	NO REACTION amides? carb. acids DO NOT react w/ NaBH_4	amide hydrolysis <chem>H3N+-CH2-COOH</chem> + <chem>H3N+-CH2-COOH</chem>	<chem>H3N+-CH2-CH2-NH-CH2OH</chem> reduces amide ? carb acid
14	 <chem>C1CCCC1C(=O)OH</chem>	NO REACTION carb acids DO NOT react w/ NaBH_4	NO REACTION carb acid hydrolyzes to carb acid!	<chem>C1CCCC1CO</chem>
15	 <chem>CH3COOCH2CH2CH(CH3)2</chem>	NO REACTION esters DO NOT react w/ NaBH_4	<chem>CH3COOH</chem> + <chem>CH3OH</chem>	<chem>CH3COOH</chem> + <chem>CH3OH</chem>
16	 <chem>C1CCCC1C(=O)N(C)C</chem>	NO REACTION amides DO NOT react w/ NaBH_4	<chem>C1CCCC1C(=O)OH</chem> + <chem>HN(C)C</chem>	<chem>C1CCCC1CN(C)C</chem>
17	 <chem>CCCC#N</chem> Ch 18 refresher	NO REACTION nitriles DO NOT react w/ NaBH_4	SKIP - reaction not covered in 8B	<chem>CCCCN</chem>

Ch 22 Rxns will NOT be included in
reaction puzzles or multi-step synthesis
on the final exam